

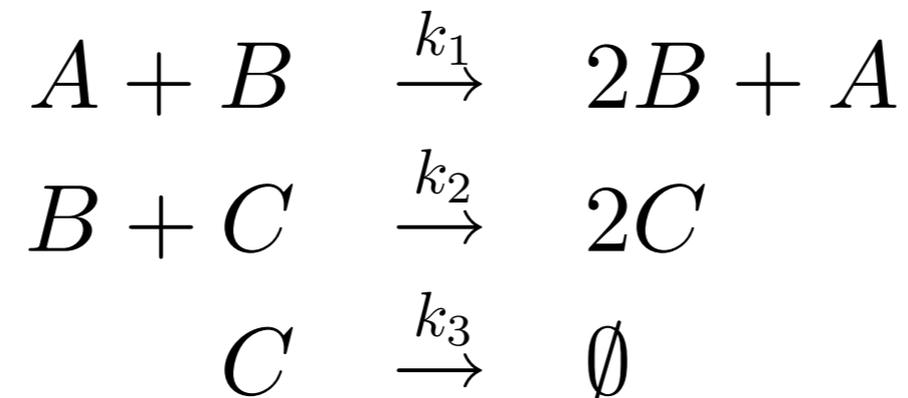
# Systems Biology 2

## Stochastic Modelling

- Any questions from the lab?

- Deterministic modeling review.
- Stochastic simulation as an alternative.
- Stochastic Simulation - Gillespie algorithm.
- Gillespie - weaknesses and extensions.

Recall the simple model:



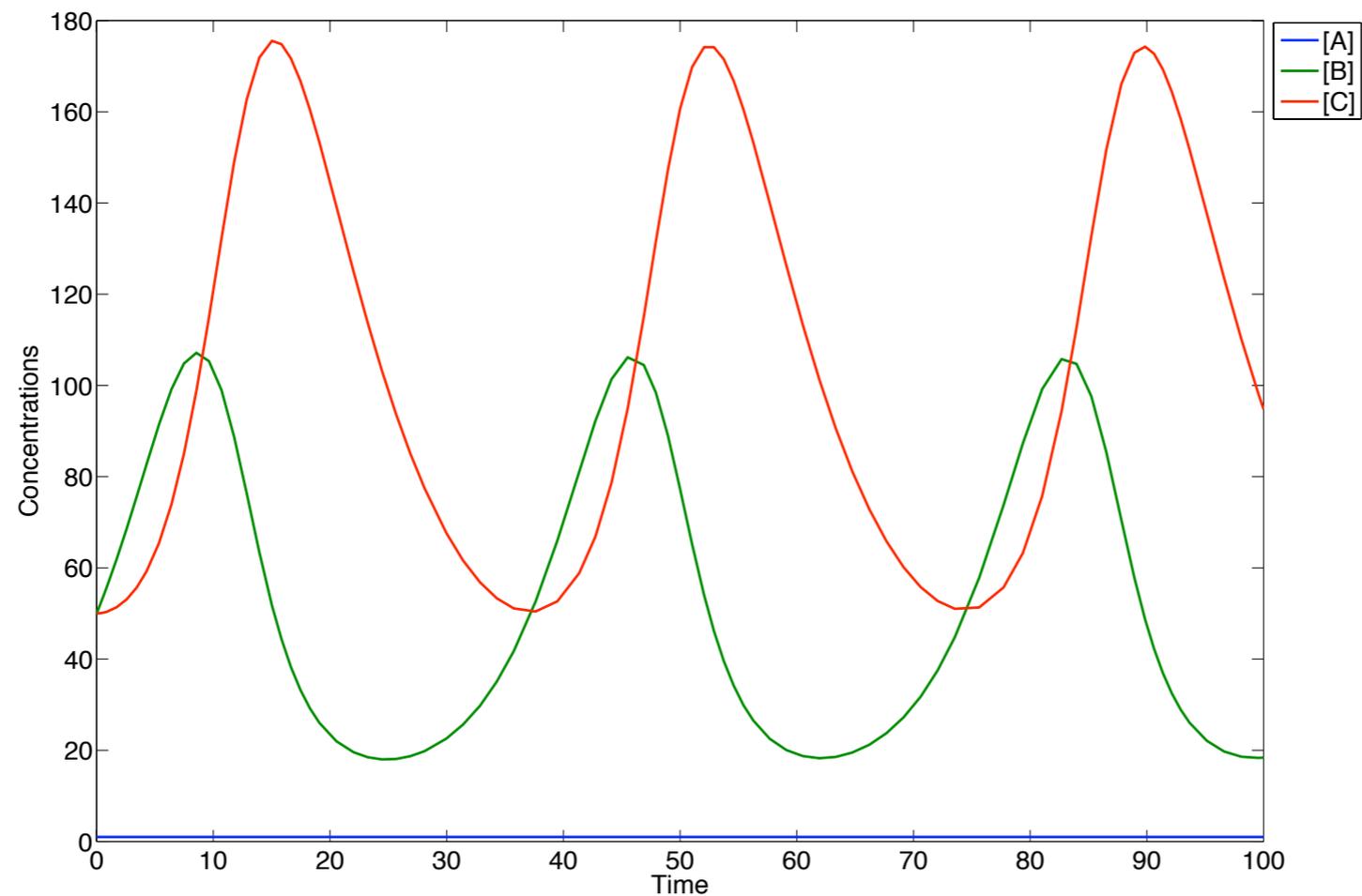
Resulting in the following set of ODEs:

$$\begin{array}{l} [\dot{A}] = 0 \\ [\dot{B}] = k_1 \cdot [A] \cdot [B] - k_2 \cdot [B] \cdot [C] \\ [\dot{C}] = k_2 \cdot [B] \cdot [C] - k_3 \cdot [C] \end{array}$$

Given initial concentrations and constants:

$$\begin{array}{lll} A|_{t=0} = 1 & B|_{t=0} = 50 & C|_{t=0} = 50 \\ k_1 = 0.25 & k_2 = 0.0025 & k_3 = 0.125 \end{array}$$

We can simulate concentrations over time:



All assumptions are wrong, some are useful...

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- In order to construct these ODEs what assumptions have we made?
  - Deterministic
  - Mass action kinetics
  - Continuous values
  - Closed system
  - Well mixed

We have assumed that we can accurately model based on some notion of average behavior.

- An alternative to ODE approaches is exact stochastic simulation.
- Exact?
  - *We deal with populations rather than concentrations.*
  - *We explicitly model each reaction.*
  - *Not: “This is exactly what will happen”!*
- *Stochastic?*
  - *Model the inherent uncertainty of the system.*
  - *Particularly important for species with small populations.*

- The most famous stochastic simulation algorithm is the Gillespie algorithm:

2340

Daniel T. Gillespie

## Exact Stochastic Simulation of Coupled Chemical Reactions

Daniel T. Gillespie\*

*Research Department, Naval Weapons Center, China Lake, California 93555 (Received May 12, 1977)*

*Publication costs assisted by the Naval Weapons Center*

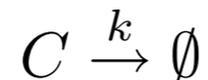
***The Journal of Physical Chemistry, Vol. 81, No. 25, 1977***

# What does Gillespie do?

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- Deals with integer populations of molecules
- Assumes the model is inherently stochastic (random)

- E.g. protein decay:



$$C|_{t=0} = 100$$

$$k = 0.1$$

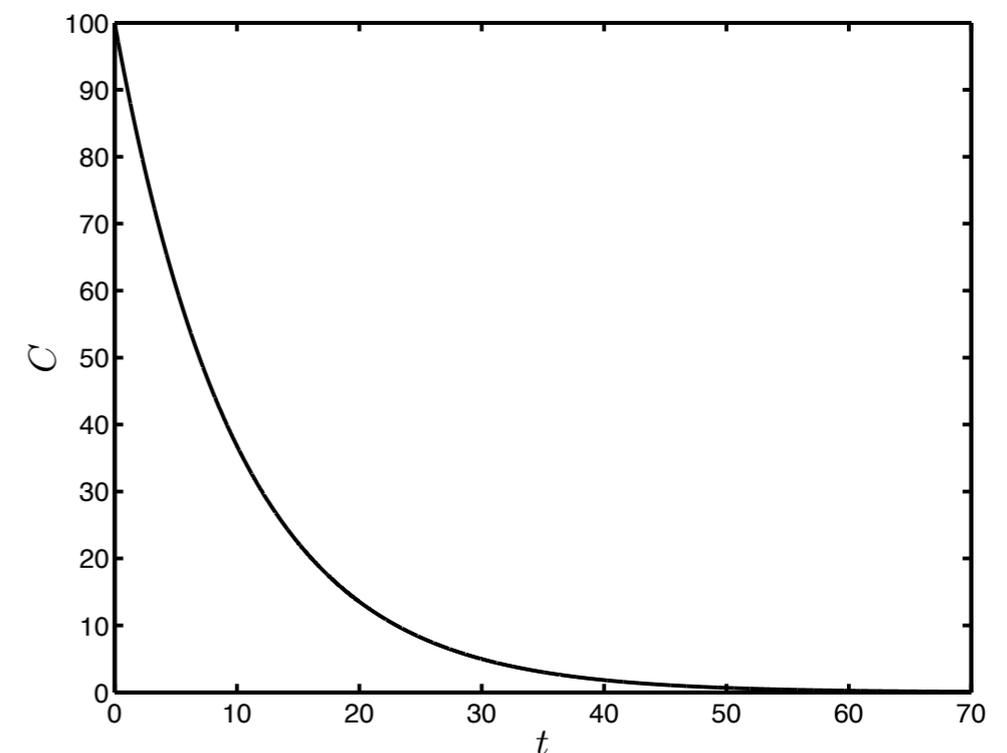
- Deterministic model (ODE) can be solved analytically

$$\dot{C} = -kC$$

$$C(t) = C_0 \exp\{-kt\}$$

And tells us that:

at  $t = 20$ ,  $C(20) = 13.5335$

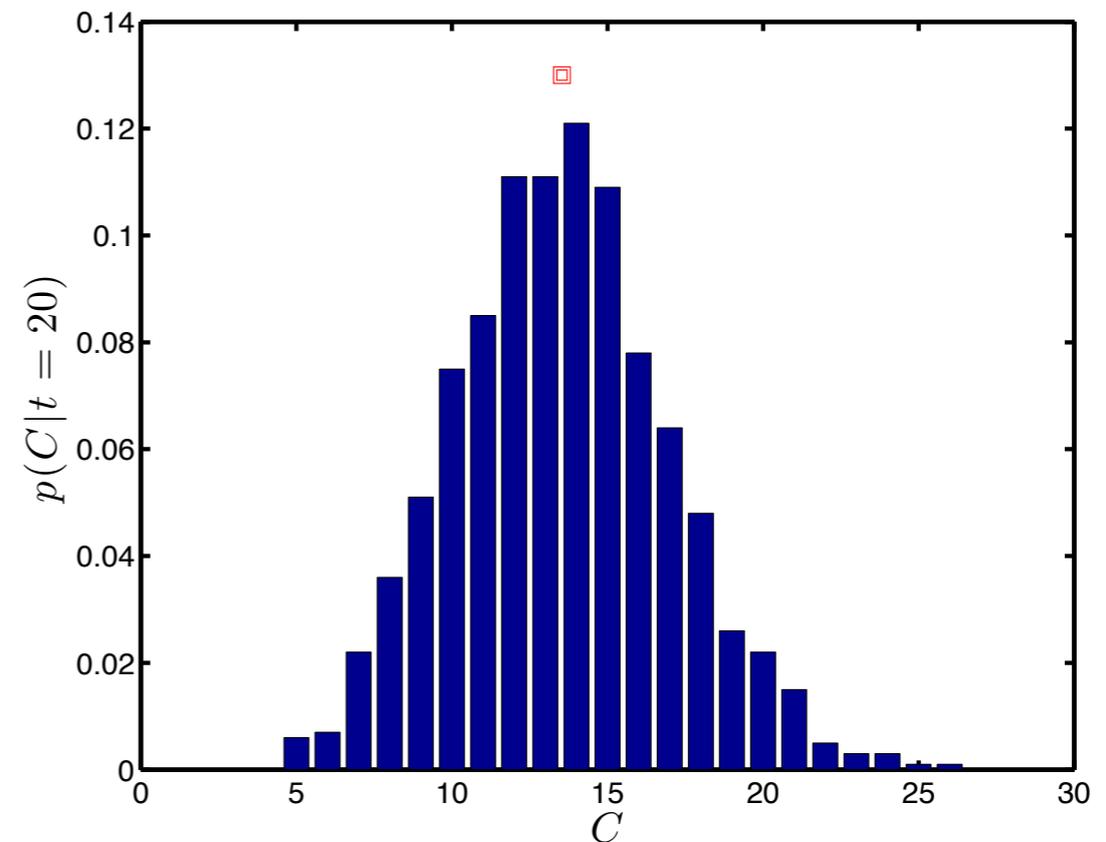


# What does Gillespie do? II

In stochastic simulation, we are interested in distributions

$$p(C|t = 20) \text{ or } p(C = 13|20)$$

Here is the distribution for  $t = 20$  created using Gillespie - we can see that the ODE (red square) in this case is in rough agreement with the most likely number of C molecules.

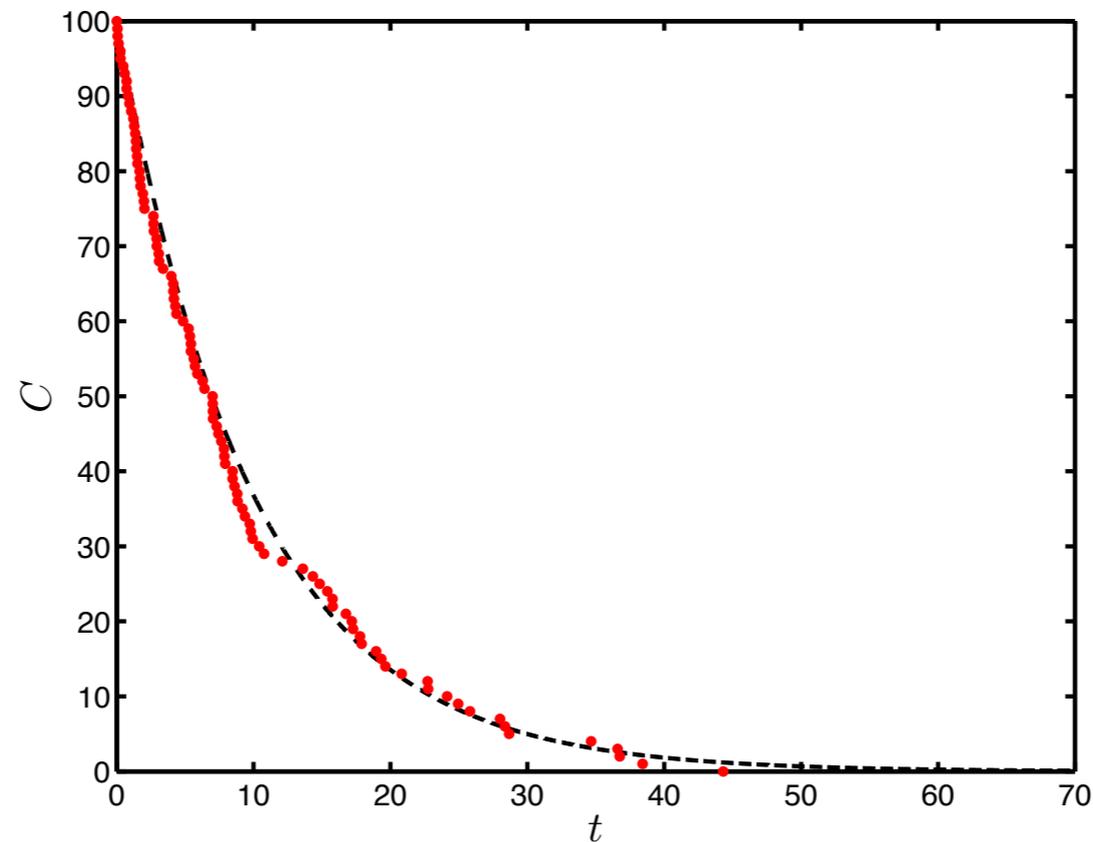


Unfortunately, for any remotely interesting model, analytically computing these probabilities is impossible. But, we can simulate...

# What does Gillespie do? III

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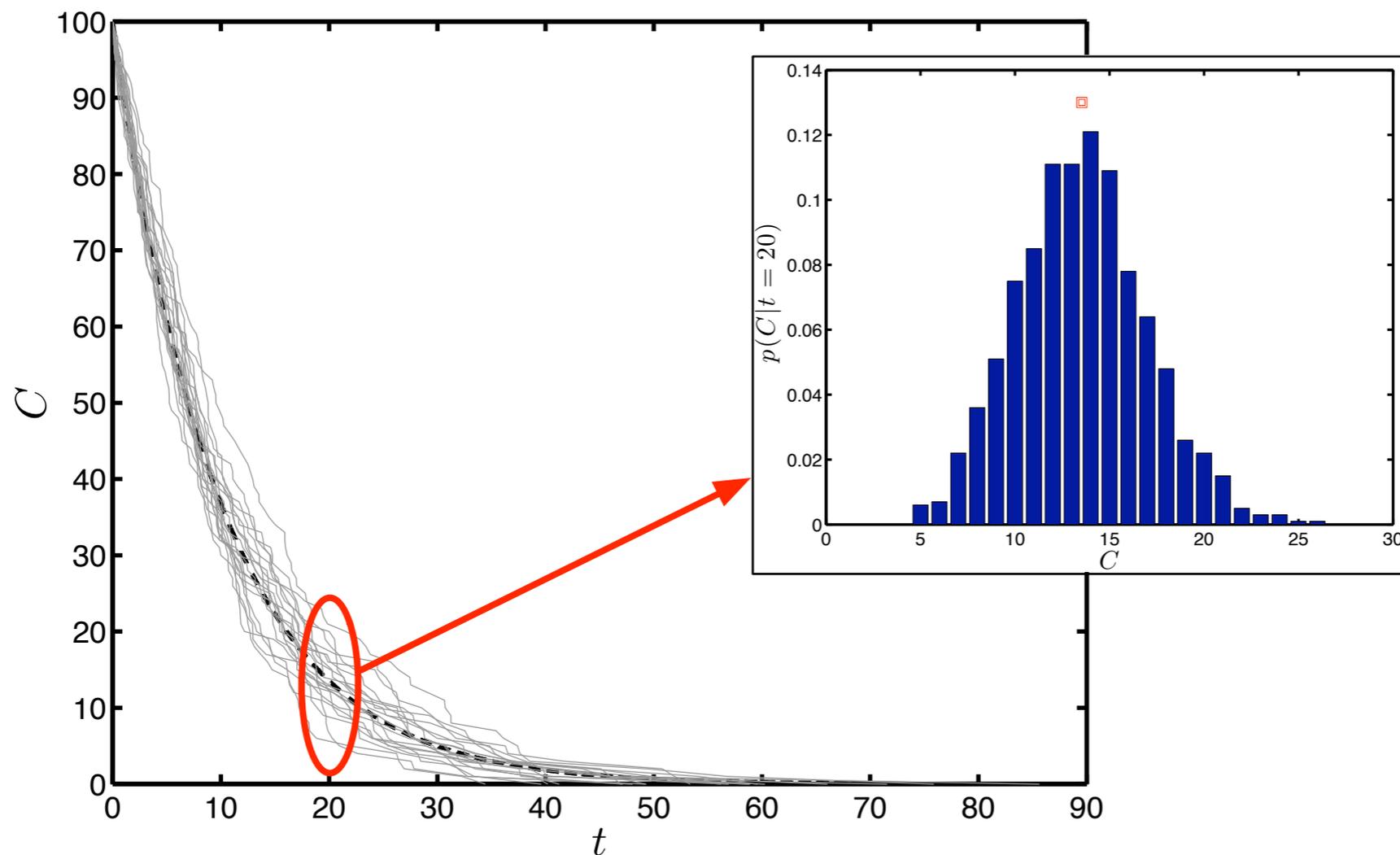
The Gillespie algorithm allows us to generate samples from the stochastic model



Each sample is a trajectory of the species' populations through time

# What does Gillespie do? IV

We may be interested in individual samples or computing empirical distributions from sets of samples...



N molecular species, M reactions

At time t, population sizes (state) given by:  $X_1, X_2, \dots, X_N$   
(Integers,  $>0$ )

**We need to generate two things:**

- 1. The time until the next reaction occurs**
- 2. The type of reaction that occurs**

Gillespie shows that the two can be de-coupled (we can sample the time and then sample which reaction takes place) resulting in a very simple procedure.

For more details and derivations, refer to Gillespie's paper.

# The Gillespie algorithm

## At each iteration...

For each of the  $M$  reactions compute:  $a_m = c_m h_m$

Reaction constant (analogous to  $k$ )



Number of reactant combinations

Reaction	Number of combination, $h$
$X_1 + X_2 \rightarrow X_3$	$X_1 X_2$
$X_1 + X_1 \rightarrow X_3$	$\frac{1}{2} X_1 (X_1 - 1)$

Then...

Let:  $a_0 = \sum_{m=1}^M a_m$

Generate:  $r_1, r_2 \sim U(0, 1)$

Compute time to reaction:  $\tau = (1/a_0) \ln(1/r_1)$

Choose reaction  $v$  for which:  $\sum_{m=1}^{v-1} a_m/a_0 < r_2 \leq \sum_{m=1}^v a_m/a_0$



$$c_1 = c_2 = 0.1$$

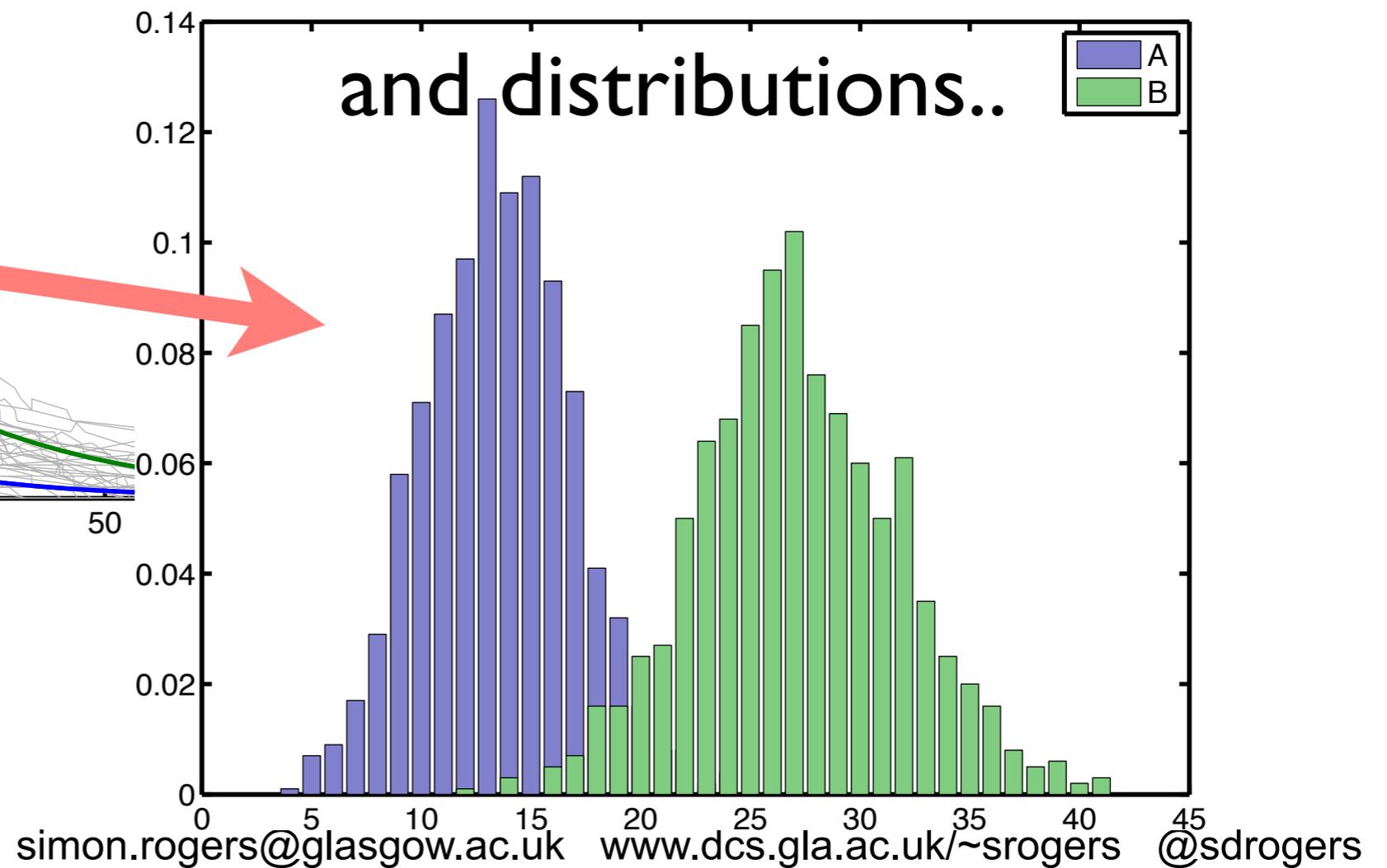
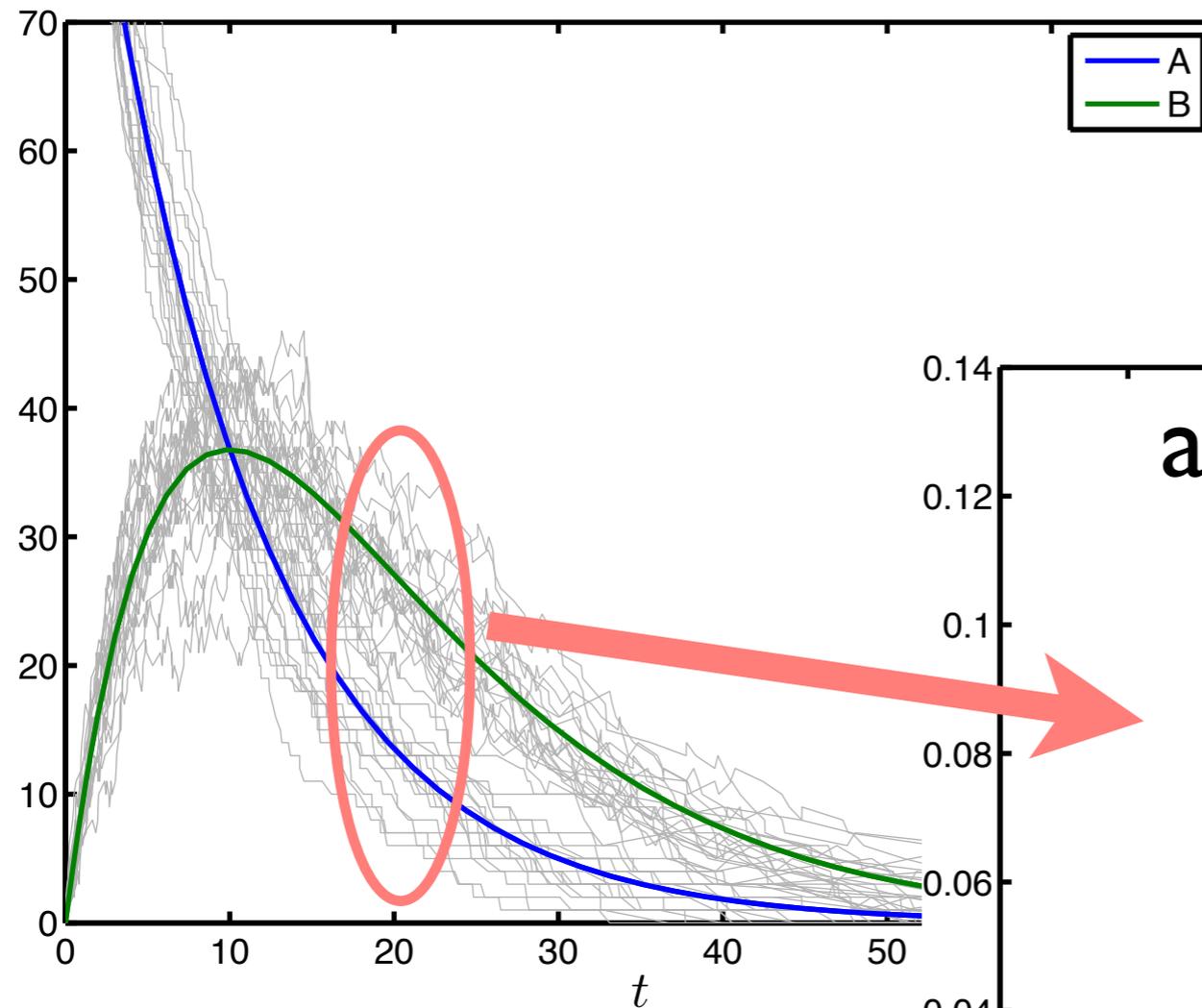
$$\text{at } t = 0, \quad A = 100, B = 0$$

# Gillespie Example

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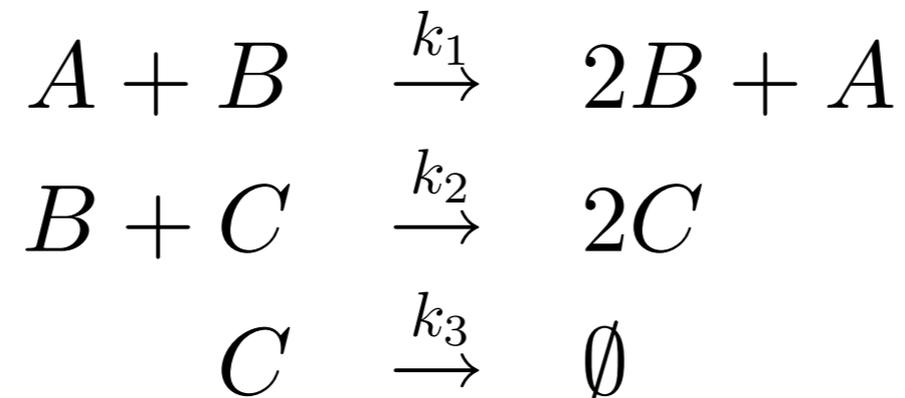
<b>It</b>	<b>t</b>	<b>A</b>	<b>B</b>	<b>h</b>	<b>a/a0</b>	<b>Sum</b>
1	0	100	0	[100,0]	[1,0]	[1,1]
2	0.4315	99	1	[99,1]	[0.99,0.01]	[0.99,1]
3	0.5528	98	2	[98,2]	[0.98,0.02]	[0.98,1]
...	...	...	...	...	...	...
32	2.8841	73	24	[73,24]	[0.75,0.25]	[0.75,1]

## Over simulations...



- As Gillespie is exact we can use it to test the assumptions we use in ODE models.
- As an example, lets test the assumptions used in the 3 species model already discussed.

Recall the simple



Resulting in the following set of  
ODEs:

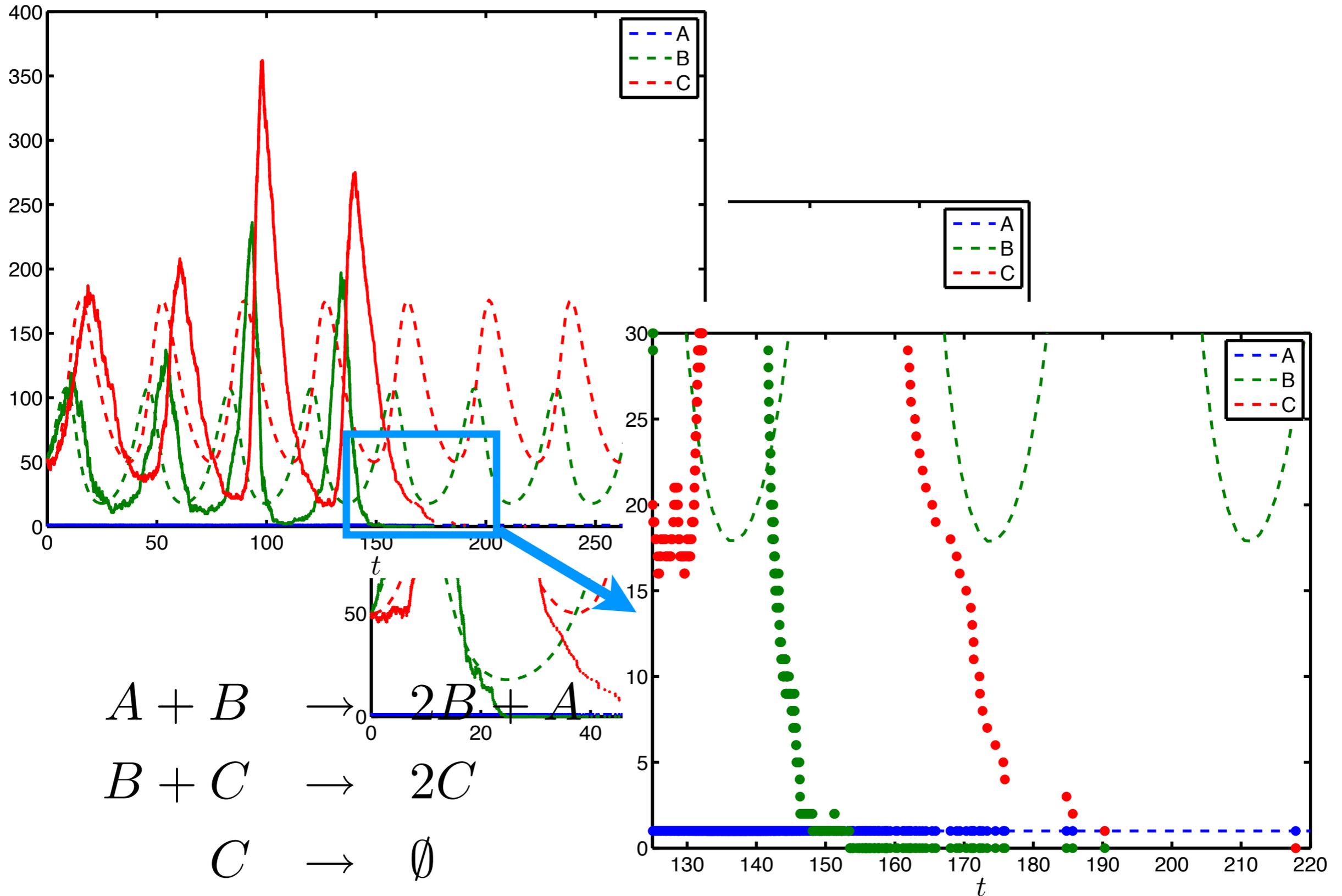
$$\begin{aligned}[\dot{A}] &= 0 \\[\dot{B}] &= k_1 \cdot [A] \cdot [B] - k_2 \cdot [B] \cdot [C] \\[\dot{C}] &= k_2 \cdot [B] \cdot [C] - k_3 \cdot [C]\end{aligned}$$

# Lotka-Volterra

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- Run the Gillespie simulator using the lotka project with  $T=100$

# Lotka-Volterra

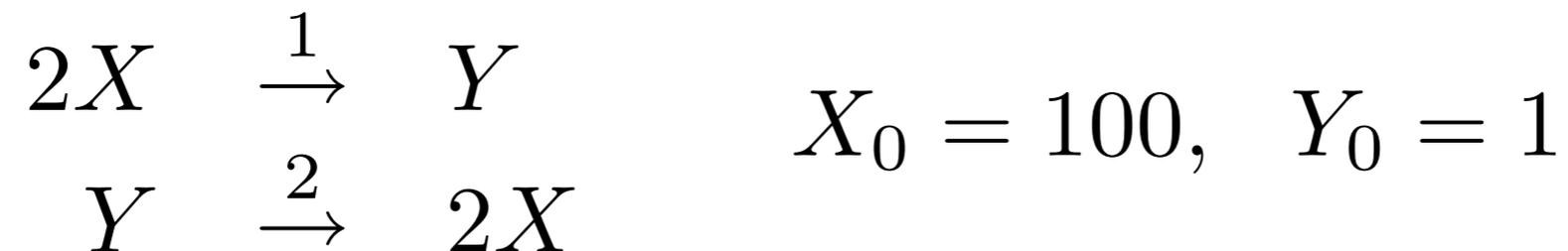


- ODE model predicts sustained oscillations ( $B > 0$ ).
- Stochastic model stops after a number of cycles ( $B = 0$ ) and never recovers.
- Very different interpretations!

# Limitations of Gillespie

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- For large populations and fast reactions, Gillespie becomes computationally impractical.
- For example, dimerisation:

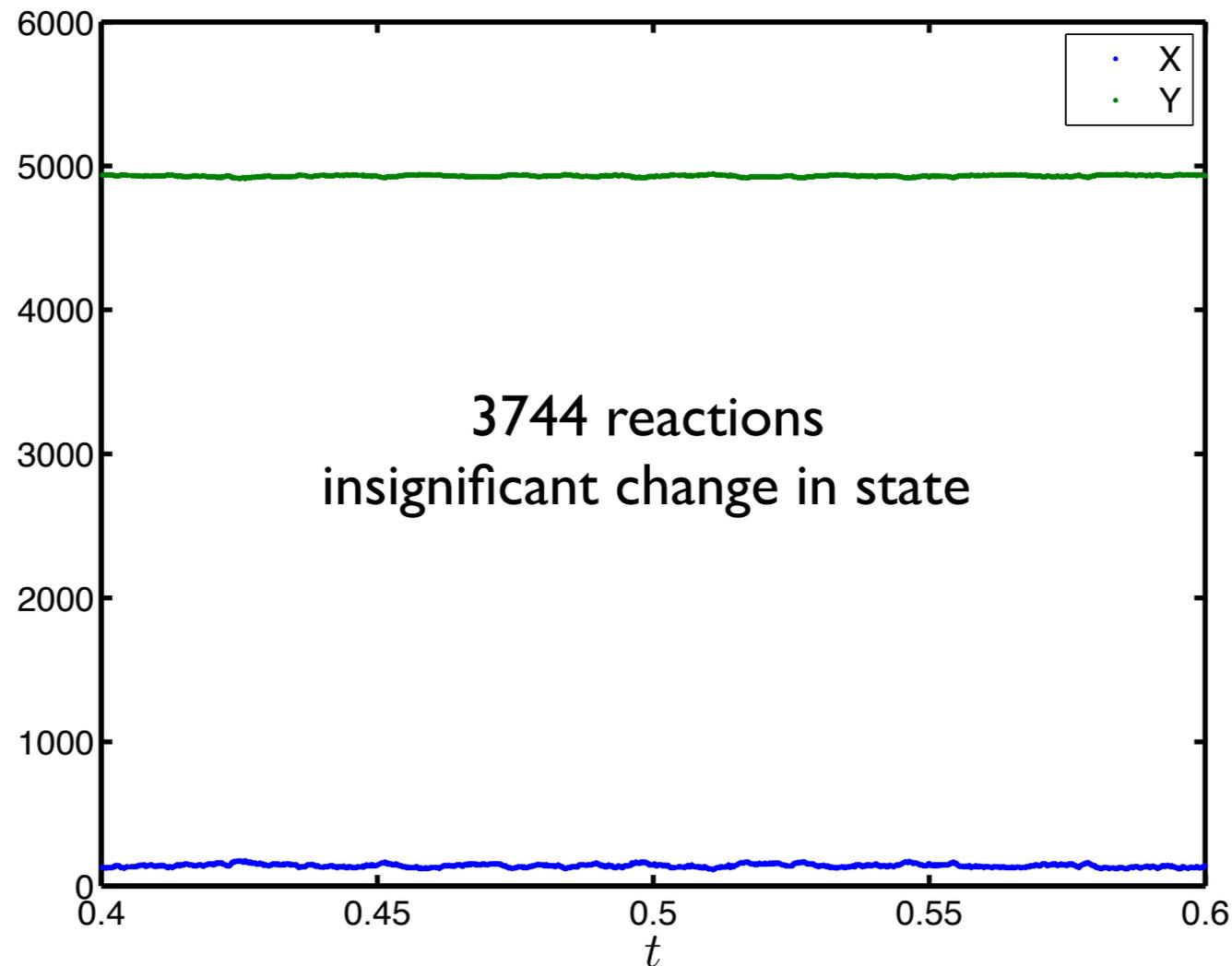


- Run this simulation (project: dimer) for  $T=100$ . How many reactions are simulated? Plot the results.

- Extensions have been proposed to speed things up:
- Exact methods:
  - Tricks to make the sampling faster.
- Approximations:
  - e.g. Tau-leaping:
    - Choose a time,  $\tau$ , to 'leap'. Compute how many reactions one would expect in that time jump.
    - Key Assumption: State does not change much in  $\tau$ .

# Extensions to Gillespie II

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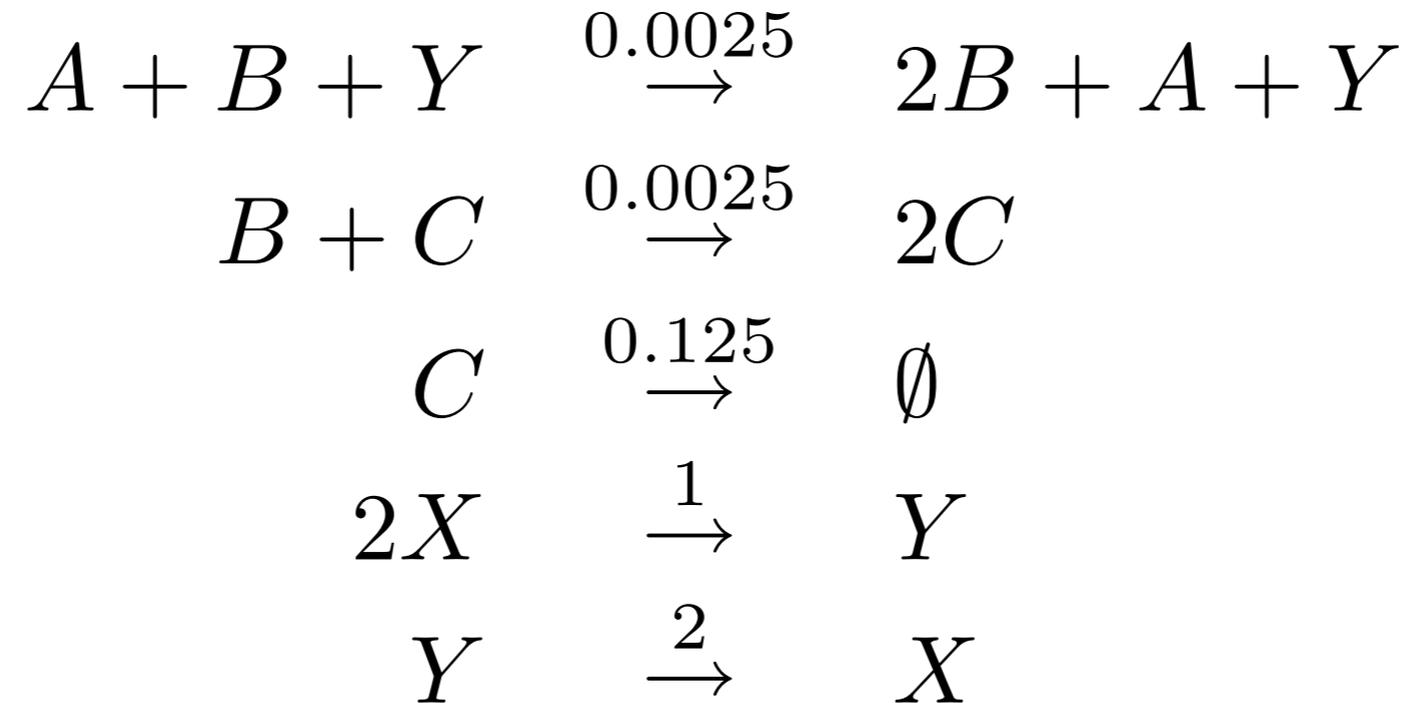


In the dimerisation example, tau-leaping may be appropriate.

- There is no reason why our model need be exclusively stochastic or deterministic.
- Mix and match:
  - Use exact for slow, important reactions
  - Use ODEs for fast, less critical reactions
- Example - Lotka-Volterra 2!

# Lotka-Volterra hybrid

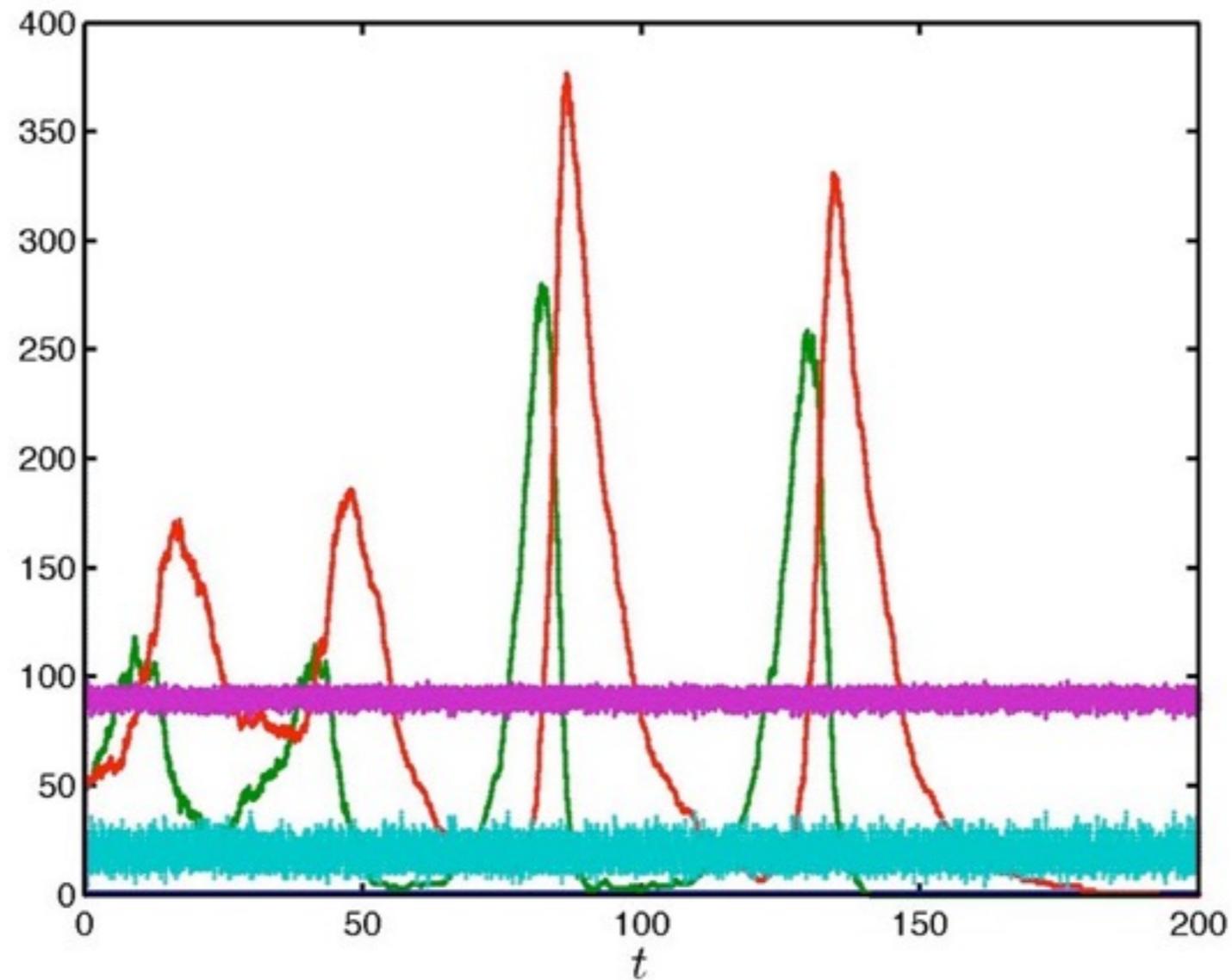
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$$A_0 = 1, \quad B_0 = 50, \quad C_0 = 50, \quad X_0 = 200, \quad Y_0 = 0$$

# Lotka-Volterra hybrid

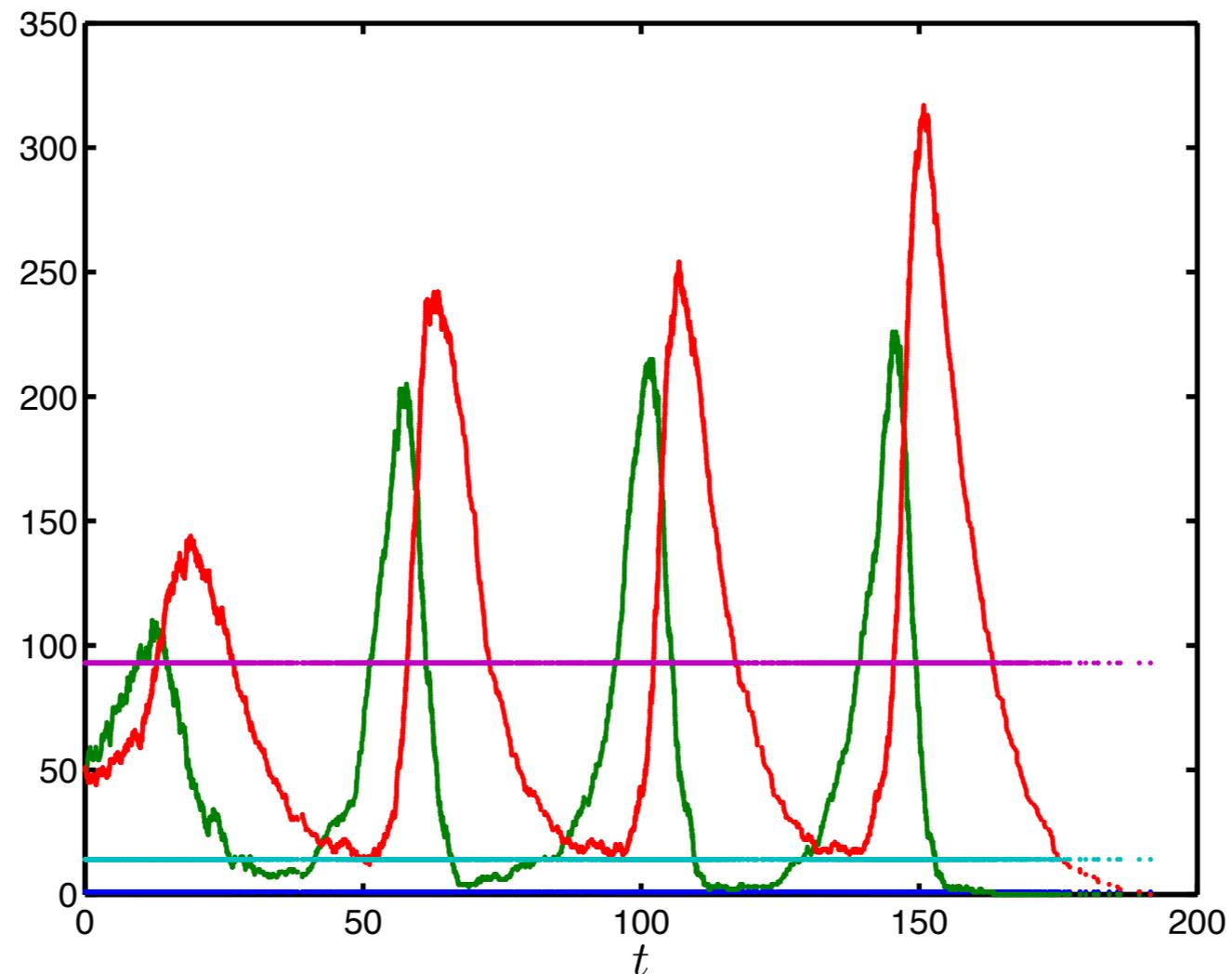
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Standard Gillespie - 78,258 reactions

# Lotka-Volterra hybrid

As  $X$  and  $Y$  only change in dimerisation reactions, remove these and substitute  $X$  and  $Y$  populations by steady state from ODEs ( $X=14$ ,  $Y=93$ ).



Same behavior - 5773 reactions (93% reduction!)

- On Wednesday, we briefly mentioned parameter estimation for ODE models
- Can the same be done with stochastic simulation?
  - Q: what kind of data?
- **Very** computationally expensive (why?)

- Exact stochastic simulation an an alternative to ODEs.
- More realistic.
- More computation!
- Can show us where ODEs breakdown.
- In large systems, stochastic simulation is not feasible. Hybrid models show great potential.